### THIN-LAYER CHROMATOGRAPHIC ANALYSIS

## OF THE PRODUCTS OF THE HYDROGENATION

OF ANABASINE, AMMODENDRINE, AND ISOAMMODENDRINE

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On reducing *l*-anabasine (I) under various conditions, Orekhov [1] obtained a mixture of hydro derivatives of anabasine from which he isolated  $l-\alpha,\beta$ -bipiperidyl. Later a number of workers [2-5] obtained this base preparatively and determined the conditions for its TLC [6].

When ammodendrine (II) was reduced and then subjected to alkaline hydrolysis, d,  $l-\alpha$ ,  $\beta$ -bipiperidyl was obtained [7]. Under similar conditions, isoammodendrine (III) forms  $d-\alpha$ , $\beta$ -bipiperidyl [8].

In hexahydroanabasine and hydrogenated N-acetylanabasine (IV) new asymmetric carbon atoms appear, which justifies the assumption of the formation of several stereoisomers from each of them. Furthermore, the bipiperidyls described in the literature [7, 8] are apparently not individual products of the reduction of saponified II and III. In view of this, in order to establish the uniformity and identify the products of the reducts of the reduction of I, II, II, and IV by comparison with authentic samples, we have used thin-layer chromatography in a nonfixed layer of alumina.

Anabasine was hydrogenated with sodium in isoamyl alcohol and also in the presence of Adams catalyst in hydrochloric acid solution. Compounds II, III, and IV were reduced only catalytically.

TABLE 1. Rf Values of the Products of the Hy-		
drogenation of $\alpha$ -Anabasine, N-Acetylanabasine,		
Ammodendrine, and Isoammodendrine		

	Rf in solvent systems		
	CHCl <sub>3</sub> (sat.	CHCH <sub>3</sub> (sat.	
	with NH <sub>3</sub> ) -	with NH <sub>3</sub> ) -	
	C.H CH.OH	C <sub>6</sub> H <sub>6</sub> — CH <sub>3</sub> OH	
	(15:18:1.5)	(35:5:1.5)	
	· · · · · · · · · · · · · · · · · · ·		
1-Anabasine	0,60	0,72	
Ammodendrine	0.58	0,77	
Dihydroammodendrine	0,44	0,66	
Isoammodendrine	0,59	0,78	
Dihydroisoammodendrine			
No. 1'	0,40	-	
No. 2'	0,44	- 1	
No. 3'	0,46		
No. 4'	0,48	l →	
α,β-Bipiperidyl			
No. 1	0,10	0,45	
No. 2	0,14	0,37	
No. 3	0,18	0,34	
No. 4	0,22	0,26	
N-Acetylanabasine	0,62	0,80	
Hexahydro-N-acetylanabasine	0.05	0.50	
No. 1"	0,25	0,53	
No. 2"	0.28	0,58	
No. 3"	0,30	0,61	
No. 4"	0,36	0,65	
Decomposition products	0.62		
No. 5	0,66		
No. 6	1 0,00	, —	

By using highly polar systems of solvents saturated with ammonia and long plates, substances with  $R_f$  values of 0.26, 0.34, 0.37, and 0.45 were detected in the products of the hydrogenation of I.

The  $l-\alpha,\beta$ -bipiperidyl obtained as described by Orekhov [1] was still nonuniform after repeated recrystallization and chromatographic purification through the hydrochloride. Two spots appeared with  $R_f$  values of two stereoisomers (Nos 2 and 3) of the four hexahydro derivatives of anabasine (Table 1).

A comparative chromatogram of the combined bases from the hydrogenated I and II shows that the saponified dihydroammodendrine (V) consists of two epimers the  $R_f$  values of which correspond to the  $R_f$ values of stereoisomers Nos. 2 and 3. In the case of V, only one spot was found.

Four substances were found in the products of the hydrogenation of III: Nos. 1' 2', 3' and 4'. The  $R_f$  values of substances No. 2 and V coincide. When the total bases from hydrogenated III were subjected to alkaline hydrolysis, substances Nos. 1, 2, 3, and 4

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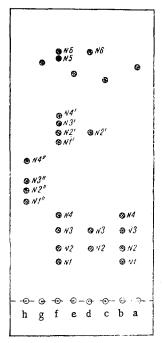


Fig. 1. Results of thinlayer chromatography: a) *l*-anabasine; b) hydrogenated *l*-anabasine; c) ammodendrine; d) hydrogenated ammodendrine; e) isoammodendrine; f) hydrogenated isoammodendrine; g) N-acetylanabasine; h) hydrogenated N-acetylanabasine. were found, the  $\mathrm{R}_{f}$  values of which correspond to those of hydrogenated I.

The combined bases from hydrogenated N-acetylanabasine contain substances Nos. 1", 2", 3", and 4", the  $R_f$  values of which do not correspond to those of V and the hydrogenated III.

Almost no products of the decomposition of anabasine were detected while for II they amounted to about 15-20% and for III to 80-85%. The R<sub>f</sub> values of these substances (Nos. 5 and 6) are greater than those of the initial II and III.

It did not prove possible to separate the stereoisomeric  $\alpha,\beta$ -bipiperidyls found and to isolate them in the pure state by preparative chromatography because of the smallness of differences in their  $R_f$  values.

## EXPERIMENTAL

*l*-Anabasine was obtained from technical anabasine sulfate via the N-acetyl derivative, bp 186-188° C/2 mm;  $n_D^{20}$  1.5516 [9]. The base was redistilled in vacuum, bp 125° C/5 mm;  $n_D^{20}$  1.5441.

Ammodendrine (from the collection of the Institute of Chemistry, Academy of Sciences of the Moldavian SSR) was the base with mp  $73-74^{\circ}$ C (from aqueous ether). Isoammodendrine (a sample given to us by Professor Kh. A. Aslanov) gave a hydriodide with mp  $218-219^{\circ}$ C.

The hydrogenation of *l*-anabasine was carried out by the action of sodium on a solution of the base in isoamyl alcohol [10]. The hydrogenation products (bp 104-117°C/2 mm; 52 g) were dissolved in 350 ml of absolute ether, and a current of  $CO_2$  precipitated a mixture of all the hexahydro derivatives in the form of carbonates, from which the combined bases were obtained by well-known methods, after which they were crystallized from a small amount of ether, mp 48-61°C. TLC showed the presence of substances Nos. 1, 2, 3, and 4. The hydrochlorides of the combined bases were precipitated from absolute ethanol and recrystallized from 95%  $C_2H_5OH$ , mp 312-314°C, base with mp 69-70°C (from ether) in which TLC showed the presence of two substances – Nos. 2 and 3.

The catalytic hydrogenation of anabasine (0.5 g), N-acetylanabasine (0.5 g), ammodendrine (0.2 g), and isoammodendrine (0.16 g) was performed with hydrogen in 0.1 N HCl solution over platinum oxide. Compounds I and IV formed mixtures of bases consisting of Nos. 1, 2, 3, and 4 and Nos. 1", 2", 3", and 4", respectively; II formed dihydroammodendrine (2'); and III formed Nos. 1', 2', 3', and 4' (bp 138-150°C/2 mm), i.e., dihydroisoammodendrine (VI). When V and VI were subjected to alkaline hydrolysis for 24 h, a mixture of substances Nos. 1 and 2 and Nos. 1, 2, 3, and 4, respectively, contaminated with the decomposition products Nos. 5 and 6, were obtained.

Thin-Layer Chromatography. The products of the hydrogenation of I, II, III, and IV and also the total bases after the hydrolysis of V and VI were dissolved in ether and chromatographed in a thin nonfixed layer of alumina (activity grade II). The solvents used were carefully purified and dehydrated methanol, benzene, and chloroform saturated with ammonia. The dimensions of the plates were  $27 \times 9$  cm and the thickness of the layer of sorbent 0.5 mm. The substances investigated were deposited on the layer of  $Al_2O_3$  in amounts of  $10-15 \ \mu g$ . These spots were revealed with iodine vapor.

### SUMMARY

It has been shown by TLC on a nonfixed layer of  $Al_2O_3$  that the hydrogenation of *l*-anabasine forms four stereoisomers and that of N-acetylanabasine three substances, and the conditions for their chromatography and their  $R_f$  values have been determined.

The  $\alpha$ , $\beta$ -bipiperidyl described previously is actually a mixture of two stereoisomers to which, according to their R<sub>f</sub> values, saponified dihydroammodendrine and dihydroisoammodendrine correspond.

The product of the reduction of ammodendrine is identical in  $R_f$  value with one of the four products from dihydroisoammodendrine. Their hydrolysis leads to the formation of the corresponding stereoisomers of  $\alpha$ ,  $\beta$ -bipiperidyl.

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